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Center for Veterinary Biologics
and
National Veterinary Services Laboratories
Testing Protocol

Supplemental Assay Method for the Determination of
Hydrogen Ion Concentration, Total Nitrogen, Phenol and
Clarity in Intradermic (Filtrate Produced From
Cultures of Pn, C, and Dt Strains of *Mycobacterium
tuberculosis*) Tuberculin

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1. Introduction

The Code of Federal Regulations, Title 9 (9 CFR) (Animals and Animal Products) states that Animal and Plant Health Inspection Service (APHIS) is responsible for administering the Virus-Serum-Toxin Act. It requires licensed tuberculin be tested and approved by the Center for Veterinary Biologics (CVB-L) before it can be marketed. Total nitrogen is determined by classical Kjeldahl digestion, distillation, and titration of the ammonia. Phenol is determined by end-point titration with bromate/bromide. The clarity is determined by visual observation of the solution; it must be clear. Satisfactory pH of the product must be 7.0 ± 0.3 . Satisfactory product must contain $0.18 \pm 0.06\%$ total nitrogen. Phenol content must be $0.54 \pm 0.04\%$.

2. Materials:

2.1 Equipment

- 2.1.1 pH meter, combination pH electrode equivalent to Orion 8103 ROSS, capable of measuring from pH 0.000 to 14.000
- 2.1.2 Balance, top loading, capable of measuring 0.01 g
- 2.1.3 Digestion unit, Buchi, B-426 with digestion tubes
- 2.1.4 Distillation unit, Buchi, B-316
- 2.1.5 Volumetric pipets, Class A, meet ASTM Standard E969-83
- 2.1.6 Volumetric flasks, Class A, with barrel head glass stopper, meet ASTM E288 Standard requirements
- 2.1.7 125 ml erlenmeyer flasks
- 2.1.8 10 ml buret with PTFE stopcock, precision bore, calibrated to ASTM E-694 accuracy requirements

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- 2.1.9 50 ml buret with PTFE stopcock, precision bore, calibrated to ASTM E-694 requirements
- 2.1.10 50, 100, 250, 500, and 1,000 ml graduated cylinders, (PYREX), meets ASTM D86, D216, and D447 requirements
- 2.1.11 Glass-stoppered 250 ml erlenmeyer flasks
- 2.1.12 Stirring plate with stirring bars
- 2.1.13 Fast filter paper, Whatman No. 1
- 2.1.14 Disposable 5 ml beaker
- 2.1.15 Rubber stopper, No. 1
- 2.1.16 Small spot light lamp

2.2 Reagents/Supplies

All chemicals, reagent grade. Use distilled or demineralized water or water of equivalent purity.

Total nitrogen

- 2.2.1 Sulfuric acid (H_2SO_4)--Purity: Minimum 95.0%
- 2.2.2 Mercury Tablets, Brinkmann Instruments, Catalog No. 015-00-646-3
- 2.2.3 Sodium hydroxide (NaOH)--Purity: 98.5%
- 2.2.4 Boric acid (H_3BO_3)--Purity: 99.9%
- 2.2.5 Methyl red--Purity: 98%
- 2.2.6 Hydrochloric acid (HCl)--Assay: 36.5-38.0%
- 2.2.7 Sodium carbonate (Na_2CO_3)--Purity: 99.9%
- 2.2.8 Bromo phenol blue--Purity: 98%

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2.2.9 National Veterinary Services Laboratories
(NVSL) Control--Pool of PPD Tuberculin products with
established protein and phenol values.

2.2.10 Protein, Bovuminar crystallized
powder--Intergen Company, Two Manhattanville Road,
Purchase, NY 10577, Catalog No. 3000-70--Purity: 98.9%

Phenol (some reagents same as for protein)

2.2.11 Methyl Orange--Purity: 98%

2.2.12 Silicotungstic Acid
($H_4[Si(W_3O_{10})_4] \cdot 26H_2O$)--Purity: 99%

2.2.13 Arsenic trioxide (As_2O_3)--Purity: 99.9%

2.2.14 Sodium Bicarbonate ($NaHCO_3$)--Purity: 99.9%

2.2.15 Potassium Bromate ($KBrO_3$)--Purity: 98.5%

2.2.16 Potassium Bromide (KBr)--Purity: 99%

2.2.17 Phenol (C_6H_5OH)--Purity: $\geq 99.0\%$

pH

2.2.18 Commercial Buffers, certified pH 7.00 and
pH 4.00

2.2.19 pH electrode storage solution, Orion, Catalog
No. 910001

2.2.20 Reference electrode filling solution, Orion,
Catalog No. 810007

3. Preparation for the Test:

3.1 Training of technical personnel

No special test related training is needed for this testing.
Analysts performing this procedure should first conduct 2
trial runs using controls and standards and obtain results
within acceptable limits.

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3.2 Preparation of equipment/instrumentation

3.2.1 pH meter (Calibrate according to the current version of GDOCSOP0009.)

Dial "Mode" to pH from stand by. Turn "Std Value" wheels to 7.000 and "Slope" wheels to 59.2 (59.2 was selected from the slope vs temperature curve provided by manufacturer). Remove the electrode from the tube of storage solution. Flush the electrode in pH 7 Buffer by means of dipping it in buffer several times and place it in second cup of fresh buffer.

Wait until the digital millivolt readout becomes stable, hit "Set Concentration" button to 7.00. For pH 4 Buffer, repeat procedure except when the digital readout becomes stable, turn "Slope" wheels until the readout is 4.000. When done, return the electrode to tube of storage solution. Check the level of reference electrode solution inside of the electrode before and after procedure.

3.2.2 Buchi Kjeldahl equipment

Become familiar with manufacturer's instructions regarding operation. Turn on water that aspirates fume from suction tube of the digestion unit and keeps the water cool in the condenser of the distillation unit. Adjust water flow in the distillation unit to approximately 1 L per minute. Turn on the distillation unit. Set time preselector to " 2 " (2 min) and stop cock for aspiration to "Off." Make sure that Buchi bottles of NaOH and water are adequate.

3.2.3 For Clarity, removal of label from the sealed bottle of tuberculin

When the bottle warms to room temperature, make sure the label is dry, and peel off the label from the bottle carefully. Clean the bottle with alcohol and lint-free paper towel.

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3.3 Preparation of reagent

3.3.1 Total Nitrogen Test (all reagents stable for at least 6 months unless specified)

1. Cut Hg tablets to one half.

Caution: Tablets contain mercury, handle in fume hood, wear gloves, safety glasses, and mask.

2. 32% NaOH, dissolve 640 ± 1 g NaOH in 1.4 L H₂O in 2 L volumetric flask on the magnetic stirrer. Cool to room temperature (RT). Dilute to volume with H₂O. Repeat above until Buchi 10 L bottle is full. Store at (RT).

Caution: NaOH is caustic, avoid contact with skin.

3. Saturated H₃BO₃, add 15 g to 100 ml H₂O. Stir, with heat, until all H₃BO₃ dissolves. Some H₃BO₃ recrystallizes when cool. Store at RT.

4. 0.5% methyl red, dissolve 0.5 g in 100 ml ethanol (95%). Store at 4°C.

5. Standardized 0.01-0.02 N HCl, 1.7 ml HCl/L H₂O. Titrate exactly 0.0100 g dried sodium carbonate dissolved in 25 ml H₂O. Indicator: 3 drops 0.1% bromo phenol blue; the color of endpoint is green not bluish green nor yellowish green. Store at RT.

Calculation:

$$N \text{ HCl} = [(g \text{ Na}_2\text{CO}_3) \times (1000)] / [(Vol \text{ HCl}) \times (52.994)].$$

Caution: Concentrated HCl is corrosive, handle in fume hood. Avoid contact with skin.

6. Protein Standard, 1 mg/ml, weigh approximately 2.00 g Bovuminar crystallized powder and transfer to 2 L flask. Dissolve and dilute to 2 L with H₂O. Aliquot into 15 ml portions in 30 ml serum vials. Seal under nitrogen. Store at 4°C.

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3.3.2 Phenol Test (all reagents stable for at least
6 mo unless specified)

1. 20% HCl, slowly add 200 ml HCl to 600 ml H₂O, dilute to 1 L. Store at RT.
2. 0.1% methyl orange, add 0.1 g methyl orange to 100 ml H₂O. Filter if necessary. Make fresh every 2 months. Store at RT.
3. Silicotungstic Acid Solution (SAS), dissolve 60 g H₄[Si(W₃O₁₀)₄]*26H₂O in 400 ml H₂O in 500 ml volumetric flask. Add 50 ml H₂SO₄. When cool, dilute to volume with H₂O. Store at 4°C.
4. Clarifying Solution (CS), add 50 ml SAS and 125 ml 20% HCl to 325 ml H₂O. Prepare fresh prior to each test.
5. "Acid Solution" for As₂O₃ standard solution, add 110 ml HCl and 2.5 ml methyl orange solution to 100 ml H₂O. Store at RT.
6. 0.0500 N As₂O₃, dissolve 2.4730 g dried As₂O₃ in 25 ml hot 1N NaOH in 1 L volumetric flask. Neutralize it with 25 ml 1N H₂SO₄. Cool and dilute to volume with H₂O. Store at RT.

Caution: As₂O₃ is extremely toxic, avoid contact, handle in fume hood using gloves, mask, and goggles.

7. Phenol Standard, dissolve 0.50 g phenol in 50 ml H₂O and dilute to 1 L. Store at RT.

Critical Control Point: The final diluted volume of the test fluid must be adjusted as described in 3.3.2.8.

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8. Test Fluid (TF), dissolve 0.30 g NaHCO_3 , 1.67 g KBrO_3 and 15.00 g KBr in H_2O and qs to 1 L with H_2O . Store at RT. The TF volume must be adjusted by adding corrected volume of H_2O to TF. It must take a volume of 21.3 ml to titrate 25 ml 0.050 N As_2O_3 in 10 ml "Acid Solution." A first time titration will require less than 21.3 ml TF. Adjust as described in the following example:

Example: Assume the first time titration volume is 20.5 ml

$$(1,000 \text{ ml of TF}) - (20.5 \text{ ml}) = 979.5 \text{ ml}$$

$$\frac{(979.5) (\text{desired vol})}{(\text{actual vol})} \text{ or } \frac{(979.5) (21.3)}{(20.5)} = 1,017.2 \text{ ml}$$

For corrected vol. of H_2O : $1017.2 - 979.4 = 37.8 \text{ ml}$ to be added to TF.

Note: TF in buret has to be put back into flask.

3.4 Preparation of the sample

3.4.1 Receipt--Reference current version of TCSOP0001.

3.4.2 Preparation

Licensed or prelicense Intradermic tuberculin products are received in sealed serum bottles. They are stored at 4°C in the walk-in refrigerator prior to testing. Before testing, allow sample vials and reagents to warm to room temperature.

4. Performance of the test: [use Tuberculin (Intradermic) Log Sheet, Appendix 9.1]

4.1 pH

Flush pH electrode in first cup of 5 ml tuberculin by means of dipping it several times till the pH readout settles, then place electrode in second cup of same fresh tuberculin. Wait until pH readout becomes stable, record pH.

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4.2 Clarity

In an area with subdued light, allow your eyes to adjust. Turn on the spot light lamp which is positioned upright. Place the unlabeled bottle over the light beam, and observe for extraneous particles.

4.3 Total nitrogen (Analyze the control pool and protein standard each time testing is performed. Analyze each in triplicate.)

4.3.1 Place 1.0 ml sample, One half Hg tablet, and 3.0 ± 0.1 ml H_2SO_4 into a Buchi Kjeldahl Tube. Same for the standard and control.

Caution: Hg is poisonous--Use gloves, mask, and goggles.

Caution: Concentrated H_2SO_4 is corrosive--Avoid contact with skin.

4.3.2 Place the tubes in Buchi digestion tube holder. Place the holder into the Buchi digestion unit. Turn on the unit and set energy regulator to "5." Fifteen min later, set to "7."

4.3.3 Digest till acid comes to true boil or no longer "burned smoke." Set to "9" for 15 more min.

4.3.4 Cool, add 6 ml H_2O , mix them and cool again.

4.3.5 Place digestion tube and a flask containing 5 ml H_3BO_3 and 3 drops indicator into the distillation unit. Tilt the flask so the tip of the condenser is immersed in the H_3BO_3 .

4.3.6 Press NaOH button and hold for $1\frac{1}{2}$ or 2 sec. Then hit Start button to start distillation unit. Distill for 2 min.

4.3.7 Titrate collected distillate to endpoint color change of yellow to deep rose (pH 5.0) with HCl. Record the volume of HCl on log sheet.

4.4 Phenol (Analyze the control pool and phenol standards each time testing is performed. Analyze each in triplicate.)

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4.4.1 Add 5 ml sample and 100 ml CS to 250 ml glass stoppered flask. Shake 2 min. Filter through fast filter paper.

4.4.2 Transfer 50 ml filtrate to another flask. Add 1 drop methyl orange, stopper, and shake a few sec. Observe the color, when red, go to 4.4.3.

4.4.3 Titrate with 2.00 ml test fluid (TF), stopper, and shake a few sec. Observe the color. When red, repeat 4.4.3. When colorless, go to 4.4.4.

4.4.4 Shake 30 sec. Add 1 drop indicator, stopper, and shake a few sec. Observe the color. When it does not turn to colorless within 10 sec, titrate with 1.00 ml TF, stopper, and repeat 4.4.4. When colorless, go to 4.4.5.

4.4.5 Shake 1 min. Add 1 drop indicator, stopper, and shake a few sec. Observe the color. When red stays longer than 10 sec, titrate with 0.50 ml TF, stopper, and repeat 4.4.5. When colorless, record total volume of TF as the endpoint of titration and use for calculation of % phenol.

5. Interpretation of the test:

5.1 pH

No calculation required.

Satisfactory pH: 7.0 ± 0.3

5.2 Clarity

No calculation required.

Satisfactory clarity: Negative (no insoluble particles observed)

5.3 Total nitrogen (Report average of triplicates.)

% Total nitrogen = $(\text{ml HCl})(N \text{ HCl})(1.4008)/(1 \text{ ml Intradermic})$.

Satisfactory total nitrogen content: $0.18\% \pm 0.06\%$

5.4 Phenol (Report average of triplicates.)

% Phenol = $(\text{volume of test fluid})(0.04) - (0.04)$

Satisfactory Phenol Content: $0.54 \pm 0.04\%$.

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5.5 Controls

Results for controls and standards must be within acceptable limits; otherwise repeat testing.

6. Report (validation) of the test results

Validate and report results according to the current version of TCSOP0001.

7. References

7.1 Code of Federal Regulations, 9 CFR, Revised January 1995, Section 113.406, page 633

7.2 Official Methods of Analysis of AOAC International, Arlington, Virginia, 16th Edition, Pat Cuniff, Editor (1995), Volume I, Chapter 12, page 7

8. Summary of Changes

Version .01 was written to meet NVSL/CVB Quality Assurance requirements, to clarify practices in use in the NVSL/CVB-L, and to provide additional detail. No significant changes were made from the previous protocol.

Version .02 was written to clarify practices in use in the NVSL/CVB-L and to provide additional detail. The following are the significant changes made from the superseded protocol:

1. Change in the digestion apparatus
2. Removal of the trichloroacetic acid precipitable protein determination step

Version .03 was written to clarify practices in use in the NVSL/CVB-L and to provide additional detail. The following is the significant change made from the previous protocol:

1. Correction of cross references in section 4

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9.1 TC Log Sheet - Tuberculin (Intradermic) Log

Tuberculin (Intradermic) Log

Cases #:	Date:
Sample Code:	Analyst:
Serial #:	Reviewed by:
Comment:	

Hydrogen Ion pH Meter	(Sample)	(Control)	Clarity (— = No insoluble particles observed, Otherwise + = Positive)
(5ml sample)	pH	pH	Sample a b c Control a

Total Nitrogen Micro-Kjeldahl		(1 ml sample)		Vol. HCl N		% Total Nitrogen		Average of 3	
Sample									
1	a								%
	b								%
	c								%
2	a								%
	b								%
	c								%
Std.:	a								%
	b								%
Control:	a								%
	b								%
	c								%

Phenol		Titration: (5ml sample)		Vol. Test Fluid		% Phenol		Average of 3	
Sample									
1	a								%
	b								%
	c								%
2	a								%
	b								%
	c								%
Std.:	a								%
	b								%
Control:	a								%
	b								%
	c								%

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10. Quick reference

- ___ Submission paperwork correct
- ___ Received correct specimens and adequate amount
- ___ Protein standard, date of origin _____
- ___ Standard acid _____
- ___ Test fluid, date of origin _____
- ___ Control Intradermic, date of origin _____
- ___ Triplicate determinations conducted
- ___ Control phenol result within limits
- ___ Control nitrogen result within limits
- ___ Nitrogen standard result within limits
- ___ Phenol standard result within limits
- ___ Results entered into computer, date _____
- ___ Results validated, date _____

Analyst _____ Date _____